Ultra-Low Fire Glass-Free Li₃FeMo₃O₁₂ Microwave Dielectric Ceramics

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A new ultra-low fire glass-free microwave dielectric material Li₃FeMo₃O₁₂ was investigated for the first time. Single phase ceramics were obtained by the conventional solid-state route after sintering at 540°C–600°C. The atomic packing fraction, FWHM of the A_g oxygen-octahedron stretching Raman mode and *Qf* values of samples sintered at different temperatures correlated well with each other. The sample with a Lower Raman shift showed a higher dielectric constant. Interestingly, the system also showed a distinct adjustable temperature coefficient of resonant frequency (from $-84 \times 10^{-6}/^{\circ}$ C to $25 \times 10^{-6}/^{\circ}$ C).

I. Introduction

THE low temperature cofired ceramic (LTCC) technology L has been used extensively for the fabrication of highperformance microelectronic packages due to its durability, compactness, and manufacturability. The LTCC technology is especially useful for microwave application. Recently, a great deal of effort has been made to fabricate low temperature sinterable microwave dielectric ceramics with a wide range of dielectric constant (ε_r), a high quality factor (*Qf*), and an adjustable temperature coefficient of resonant frequency (τ_f) .¹ However, at present reducing the sintering temperature without affecting the electrical performance is still a challenge. The commonly used methods to reduce the sintering temperature, including low melting glass addition, chemical processing, and use of fine raw materials,^{2,3} usually lead to complicated fabrication procedure and deteriorated properties. Recent studies showed that the compound with lowmelting temperature oxides, such as TeO₂ (733°C), MoO₃ (795°C), Bi₂O₃ (817°C), and Li₂CO₃ (723°C). etc., could give rise to a low or even ultra-low sintering temperature.⁴ While without glass addition, it is still hard to lower the sintering temperature below 900°C. Exceptions are the molybdate and tellurate, which can be densified below 750°C.^{6–10} Tellurate, like $Zn_3Te_3O_8$ -TiTe₃O₈, CaTe₂O₅-TeO₂ and SrTe₂O₅-TeO₂, were reported to have a sintering temperature in the range of about 600°C-650°C. Nevertheless, they still show good electric properties with a dielectric constant ε_r of about 19.8–28.7, *a Qf* value in the range of 12 000– 50 000 GHz, and a τ_f in the range of about -3.8×10^{-6} /°C -3.0×10^{-6} /°C.^{11,12} However, TeO₂ has the problems of high cost and toxicity. In contrast, molybdate is much more advantageous for practical use in LTCC microwave components. Typical molybdate like $Li_2(M^{2+})_2Mo_3O_{12}$ and $Li_3(M^{3+})Mo_3O_{12}$ (M=Zn, Ca, Al, and In) were reported to have relatively high Qf values (36 000~70 000 GHz) and a

low sintering temperature (570°C~630°C), but their τ_{f} is still large (-90 \times 10⁻⁶/°C \sim -73 \times 10⁻⁶/°C).⁶ Here, we report a new microwave dielectric material Li₃(M³⁺)Mo₃O₁₂ (M=Fe) that have ultra-low sintering temperature (540°C~600°C) with adjustable τ_{f} (-84 \times 10⁻⁶/°C \sim 24 \times 10⁻⁶/°C). The crystal structure, Raman scattering spectra, and dielectric properties of the material were investigated.

II. Experimental Procedure

The Li₃FeMo₃O₁₂ ceramics were prepared by the solid-state method. High-purity (>99.9%) Li₂CO₃, Fe₂O₃ and MoO₃ raw materials were ball-milled using de-ionized water as the medium for 10 h. The dried powders were calcined (500°C for 3 h), ground, and pressed into pellets. After sintering at 540°C~600°C for 6 h, ceramic samples with a diameter of 10 mm and a thickness of 5 mm were obtained. The phase structure was examined by XRD (D8; Brucker, Stuttgart, Germany). The morphology of the fractured surfaces of the samples was examined by a Scanning Electron Microscope (S-4800; Hitachi, Tokyo, Japan). Microwave dielectric properties were measured by a network analyzer (8720ES; Agilent, Santa Clara, CA) in the frequency range of 12.0~14.5 GHz. The dielectric constant was measured by the Hakki-Coleman method, as modified by Courtney,13 and the unloaded Q values were measured by the cavity method.¹⁴ The τ_f values were determined from the resonant frequencies in the temperatures range of 25°C-85°C. The measurement error of dielectric constant, τ_f and Qf were less than 0.5% $\times \varepsilon_r$, 0.5 \times 10⁻⁶/°C $\times \tau_f$ and 5% $\times Qf$, respectively.

III. Results and Discussion

The XRD patterns of the Li₃FeMo₃O₁₂ samples shown in Fig. 1 indicate that single orthorhombic phase structure (ICSD: 16176) was obtained in all the temperatures. Such a structure belongs to the space group Pnma (D_{2h}^{16}) and contains four Li₃FeMo₃O₁₂ molecules per primitive cell. From the schematic illustration of the Li₃FeMo₃O₁₂ supercell $(4 \times 4 \times 4)$ in the inset of Fig. 1, we can see that the Li₃FeMo₃O₁₂ lattice consists of edge/corner shared Li/FeO₆ octahedra and MoO₄ tetrahedra. Each unit contains one central Li/FeO₆ octahedron surrounded by six corner-shared MoO₄ tetrahedra, which are in turn connected with twelve edge-shared peripheral Li/Fe octahedra. The lattice parameters calculated from the XRD patterns are listed in Table I, from which the theoretical density¹⁵ and packing fraction can be calculated. As shown in Fig. 2, the density increases with increasing sintering temperature. The relatively density (> 95%) indicates that the Li₃FeMo₃O₁₂ ceramics can be densified at 540°C. The inset of Fig. 2 is a SEM image of a typical sample (600°C), which consists of rod-shaped grains with a diameter of 0.5~1 µm and a length of 2~2.5 µm. Other samples have a similar morphology.

Figure 3 Shows the Raman spectra of different $Li_3FeMo_3O_{12}$ samples. The material has 76 atoms in its unit cell (see the inset of Fig. 1), which give rise to 228 spectrum

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 Table I.
 Crystallographic Data of Li₃FeMo₃O₁₂ Microwave Deleictric Ceramics

Cintania -	Lattice parameters (Å)			
temperature (°C)	а	b	с	Vol (Å ³)
540	17.58797	10.48384	5.09483	939.43
560	17.64741	10.4613	5.08227	938.26
580 600	17.57049 17.57880	10.47891 10.47976	5.09788 5.09721	938.62 939.02



Fig. 1. The XRD patterns of the $Li_3FeMo_3O_{12}$ ceramics after sintering at 540°C ~ 600°C. The inset shows the (4 × 4 × 4) supercell of the lattice, where the blue and yellow parts represent Li/ FeO₆ octahedron and MoO₄ tetrahedron, respectively.



Fig. 2. The bulk density and relative density of the $Li_3FeMo_3O_{12}$ ceramics as a function of sintering temperature. The inset shows the SEM photograph of a typical sample that has been sintered at 600°C.

branches. Using the group theoretical method,¹⁶ and via the irreducible representations for the lattice, the Raman active modes can be calculated as $\Gamma_{raman}=5A_g+4B_1 \ _g+5B_2 \ _g+4B_3 \ _g$ (Γ_{raman} is the irreducible representation). Previous studies have shown that the A_g mode scattering, which is related to the stretching vibrations of the oxygen octahedron, are closely related to the microwave dielectric properties.^{17,18} The A_g mode scattering frequency and strong intensity.^{17,18} Therefore, the strongest peak around 933 cm⁻¹ is assigned to the A_g mode scattering, which will be discussed in the following.



Fig. 3. The Raman scattering spectra of the $Li_3FeMo_3O_{12}$ ceramics after sintering at 540°C ~ 600°C. The main Raman active vibrational modes are marked by the dash lines.



Fig. 4. The Qf value of the Li₃FeMo₃O₁₂ ceramics as a function of sintering temperature. The inset shows the packing fraction and 1/FWHM of the A_g mode Raman scattering peak of different ceramic samples.

As an important structural parameter, the packing fraction (the summation of the volume of packed ions over the volume of a primitive unit cell) has a strong influence on the dielectric properties, which could be obtained from Eq. (1):¹⁹

packing fraction (%) =
$$\frac{\text{volume of the atoms in the cell}}{\text{volume of primitive unit cell}}$$

= $\frac{\text{volume of the atoms in the cell}}{\text{volume of unit cell}} \times Z$
(1)

where Z is the number of formula units per unit cell. As shown in the inset of Fig. 4, the packing fraction increases from 61.5930% to 61.6697% when the sintering temperature is increased from 540° C to 560° C, and then it decreases to 61.6198% when the temperature is further increased to 600° C.

The inverse of the FWHM (Full Width at Half Maximum) of the A_g mode of the Raman peak, the packing fraction, and *Qf* values of different Li₃FeMo₃O₁₂ ceramic samples are shown in Fig. 4. A rather good correlation among the data can be clearly seen. Namely, all three parameters reach maximum values at 560°C and then decreases slightly with further

13.2

12.6

->- Dielectric constant &

Temperature coefficient of

– Raman shift

932.0

932.5

40

20



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Fig. 5. The dielectric constant ϵ , the A_g mode Raman shift and the temperature coefficient of resonant frequency τ_f of different Li₃FeMo₃O₁₂ ceramics samples.

increasing of temperature. The *Qf* value is related to the dielectric loss from anharmonic vibrations of the atoms.²⁰ The increase in packing fraction means a decreased vibration space, and this in turn may give rise to a decrease in anharmonic vibration. Consequently, a good correlation between Of and packing fraction is observed. The decrease in the 1/FWHM of the A_g mode reflects a decreased anharmonici-ty. The anharmonicity of the vibration is related to the pho-non damping²¹ and in turn to the intrinsic dielectric loss. Therefore, the narrowing of FWHM of the A_g mode peak implies a lager damping and dielectric loss, and hence a lower Of value is observed.

The correlation between Raman shift and the dielectric constant ε_r as well as the temperature coefficient of resonant frequency τ_f are shown in Fig. 5. A lower Raman shift implies a more rigid oxygen octahedron and a lower polarizability, and this in turn results in a lower ε_r . Therefore, the Raman shift and ε_r show a good correlation. An interesting phenomenon reflected from Fig. 5 is that, with the increase in the sintering temperature, the τ_f value varies almost linearly from -84.5912×10^{-6} /°C to 24.8967×10^{-6} /°C. This implies that the new Li₃FeMo₃O₁₂ system has distinct adjustable microwave performance. This character may be used to meet specific requirements of a certain devices and can be achieved by controlling the sintering temperature. This is probably related to the specific crystal structure of Li₃FeMo₃O₁₂. The underlying mechanism deserves further investigation.

IV. Conclusions

The crystal structure, Raman scattering spectra and microwave dielectric properties of a new Li₃FeMo₃O₁₂ microwave dielectric material were investigated. The results showed that the Li₃FeMo₃O₁₂ ceramics could be densified at 540°C. The packing fraction, FWHM of the Raman Ag mode and Qf value of samples sintered at different temperatures correlated well with each other. Also, the sample with a lower Raman shift showed a higher dielectric constant. The new system also exhibits a distinct adjustable temperature coefficient of resonant frequency that can be controlled by sintering temperature.